

Announcements

- HW3: 13, 17, 23, 25, 28, 31, 37, 38

*** Course Web Page **

<http://highenergy.phys.ttu.edu/~slee/2402/>

Lecture Notes, HW Assignments,
Schedule for the Physics Colloquium, etc..

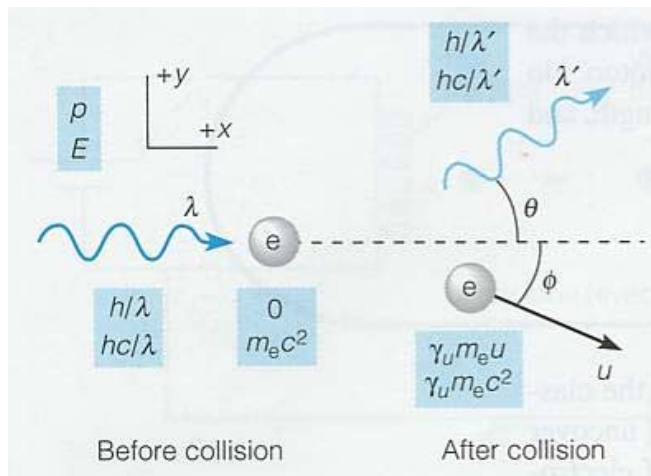
Lecture 8 – Chapter. 3 Wave & Particles I

EM-“Waves” behaving like “Particles”

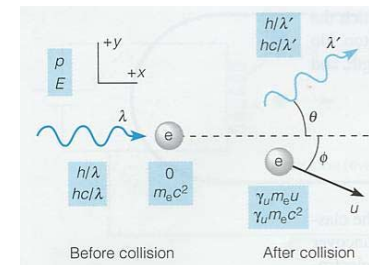
Outline:

- Blackbody Radiation (Plank; 1900; 1918*)
- The Photoelectric Effect (Einstein; 1905; 1921*)
- The Production of X-Rays (Rontgen; 1901; 1901*)
- The Compton Effect (Compton; 1927; 1927*)
- Pair Production (Anderson; 1932; 1936*)
- Is It a Wave or a Particle? → Duality?

Momentum & Energy when a photon strike a free electron



Energy and Momentum Conservation



Momentum conserved:

$$x\text{-component: } \frac{h}{\lambda} = \frac{h}{\lambda'} \cos \theta + \gamma_u m_e u \cos \phi$$

$$y\text{-component: } 0 = \frac{h}{\lambda'} \sin \theta - \gamma_u m_e u \sin \phi$$

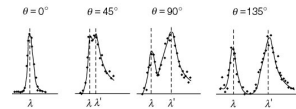
Energy conserved:

$$h \frac{c}{\lambda} + m_e c^2 = h \frac{c}{\lambda'} + \gamma_u m_e c^2$$

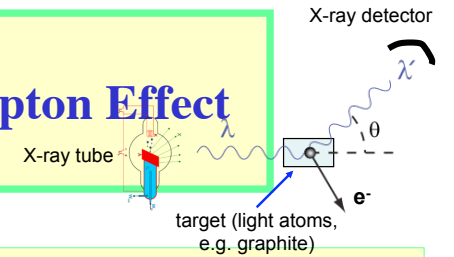
The photoelectric effect and the Compton effect are two important ways in which EM radiation interacts as a particle with matter.

We now discuss a third!

$$\lambda' - \lambda = \frac{h}{m_e c} (1 - \cos \theta)$$



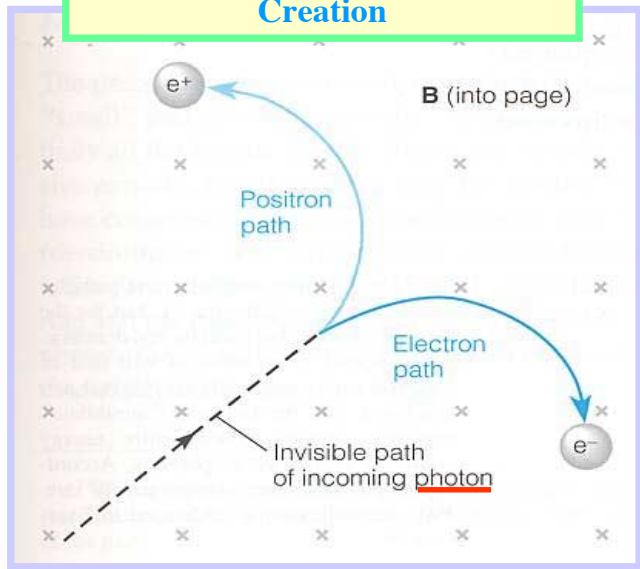
The Compton Effect



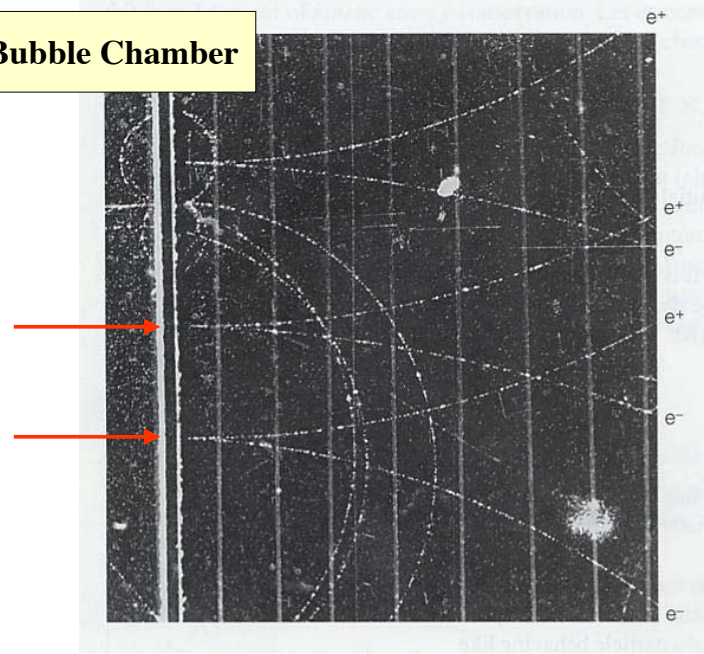
Photons carry momentum like particles and scatter individually with other particles

indeed, the wavelength shift is independent of the target material and the initial photon wavelength.

Particle-Antiparticle Pair Creation



Bubble Chamber



Example 2.5

Calculate the energy and wavelength of the least-energetic photon capable of producing an electron-positron pair.

Solution

The energy in the photon becomes the energy of the massive particles, internal/mass energy plus any kinetic energy. The minimum energy required is that which is barely able to produce the pair, with no kinetic energy. In this case, the photon energy equals just the internal energy of the pair:

$$\begin{aligned} 2 \cdot m_e c^2 &= 2(9.11 \times 10^{-31} \text{ kg})(3 \times 10^8 \text{ m/s})^2 \\ &= 1.64 \times 10^{-13} \text{ J} \quad (\approx 1 \text{ MeV}) \end{aligned}$$

Thus,

$$\begin{aligned} \frac{h c}{\lambda} = 2 m_e c^2 &\Rightarrow \lambda = \frac{h c}{2 m_e c^2} \\ &= \frac{(6.63 \times 10^{-34} \text{ J}\cdot\text{s})(3 \times 10^8 \text{ m/s})}{1.64 \times 10^{-13} \text{ J}} \\ &= 1.21 \times 10^{-12} \text{ m} \end{aligned}$$

Electromagnetic Waves

behaving like

Particles “PHOTONS”

(Chapter 2)

Black Body Radiation
The Photoelectric Effect
The Production of X-Rays

PHOTONS
 $E = hf$

The Compton Effect

PHOTONS
 $p = hf/c = h/\lambda$

Particle-Antiparticle Pair Production

Photon properties:

$$E = hf \quad (2-1)$$

$$p = \frac{h}{\lambda}$$

Photoelectric effect:

$$KE_{\max} = hf - \phi \quad (2-2)$$

Compton effect:

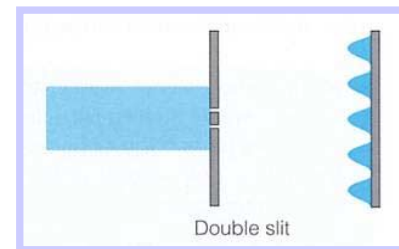
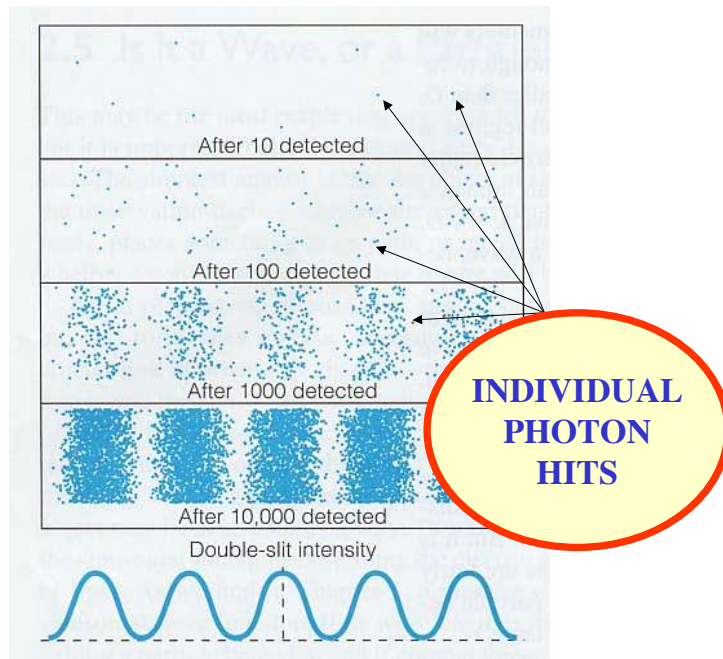
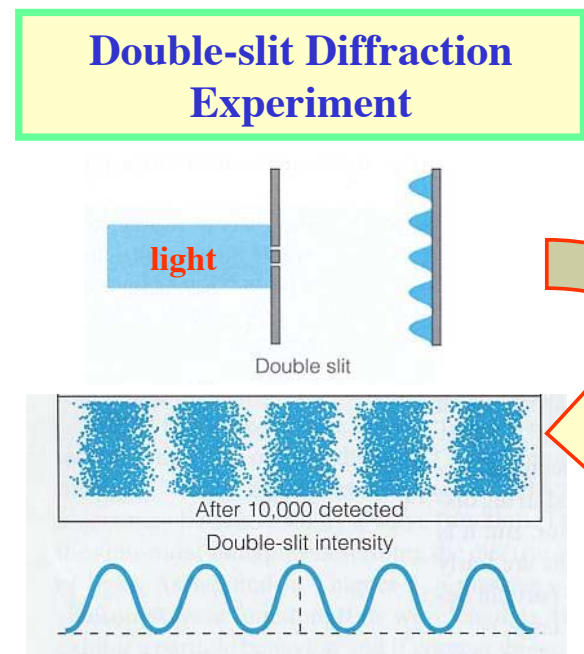
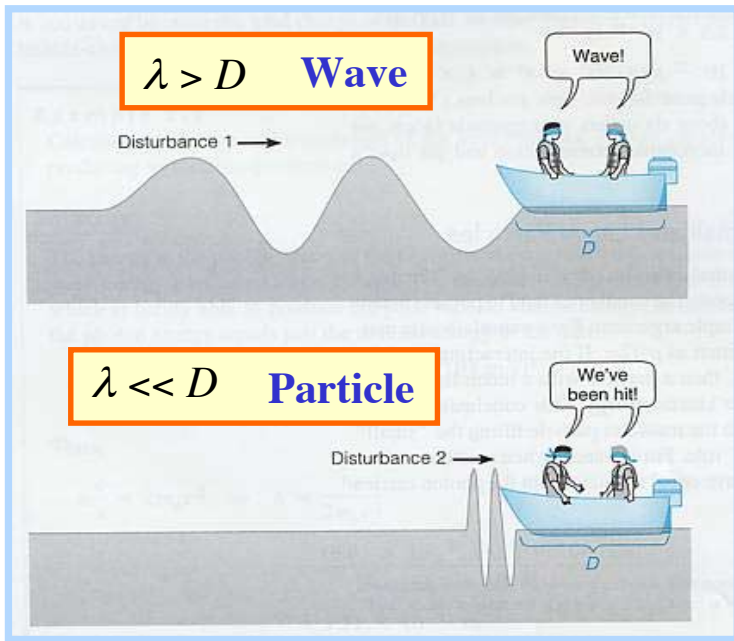
$$\lambda' - \lambda = \frac{h}{m_e c} (1 - \cos \theta)$$

Is It a Wave or a Particles? Duality

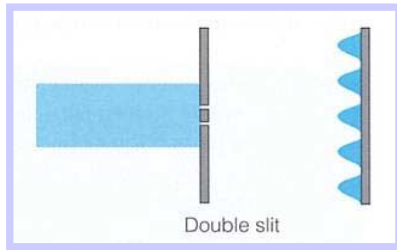
Ch.2:: EM-Waves behaving like Particles
Ch.3:: Particles behaving like Waves

Outline:

- Particles or Waves
- A Double-Slit Experiment – Light Interference
- A Double-Slit Experiment – Matter Wave Interference
- Properties of Matter Waves
- **Uncertainty Principle**



Although diffraction of light is a *wave phenomenon*, there is no smooth distribution of light in the diffraction pattern, but the pattern is rather formed of many *individual hits of particles – the photons*



**A single photon
DOES NOT
get “disintegrated” in the
Diffraction process
to make a smooth
diffraction pattern**

When a phenomenon is detected as *particles*, it cannot be predicted with certainty where a given particle will be found. The most that can be determined is a probability of finding it in a given region, and this “probability density” is proportional to the square of the amplitude of the associated *wave* in that region.

$$\text{probability density of finding } \textit{particle} \propto (\text{amplitude of } \textit{wave})^2$$

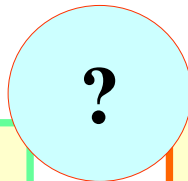
Coming back to that soon...

Electromagnetic
waves (light)



Particles
(photons)

Waves



Massive
Particles